## **5 Physics 2424 Atoms**

1. The probability of spontaneous transitions of a charged particle from one initial energy level to another final level is governed by the overlap integral

$$
\vec{D}_o = q \int \vec{r} \psi_i^* \psi_f dV,
$$

where *q* is the charge of the particle. If  $\vec{D}_o = 0$  between two levels, the transition is forbidden. Searching for such forbidden transitions leads to *selection rules*. Consider the 3D infinite potential well.

- (a) Show that only allowed transitions from  $\psi_{n1,n2,n3}$  are to a state where two of the quantum numbers are unchanged and the other quantum number changes by an odd number like 1, 3, 5, etc.
- (b) Show that the emitted photons have energies  $E = p(2n p)E_1$  where the *n* is the original quantum level and  $n - p$  is the final quantum level and  $E_1$  is the ground state energy of the one-dimensional potential well.

The following integral identities are useful:

$$
\int_0^L \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi x}{L}\right) dx = \frac{L}{2} \delta_{nm} \text{ and } \int_0^L x \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi x}{L}\right) dx = \begin{cases} \frac{L^2}{4} & n = m \\ \frac{2L^2}{\pi^2} \frac{n m \left((-1)^{n-m} - 1\right)}{\left(n - m\right)^2 \left(n + m\right)^2} & n \neq m \end{cases}
$$

The wavefunction for the symmetric 3D potential is given by

$$
\psi(x, y, z) = \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi y}{L}\right) \sin\left(\frac{o\pi z}{L}\right).
$$

Also  $\vec{r} = i x + j y + k z$ . The overlap integral therefore is

$$
\vec{D}_o = e \int_0^L \int_0^L \left( i x + j y + k z \right) \sin \left( \frac{n_i \pi x}{L} \right) \sin \left( \frac{m_i \pi y}{L} \right) \sin \left( \frac{o_i \pi z}{L} \right) \sin \left( \frac{n_j \pi x}{L} \right) \sin \left( \frac{m_j \pi y}{L} \right) \sin \left( \frac{o_j \pi z}{L} \right) dx dy dz
$$

Considering the *x* component we find

$$
D_{ox} = e \int_0^L x \sin\left(\frac{n_i \pi x}{L}\right) \sin\left(\frac{n_f \pi x}{L}\right) dx \int_0^L \sin\left(\frac{m_i \pi y}{L}\right) \sin\left(\frac{m_f \pi y}{L}\right) dy \int_0^L \sin\left(\frac{o_i \pi z}{L}\right) \sin\left(\frac{o_j \pi z}{L}\right) dz
$$

The *y* and *z* integral will give zero unless  $m_f = m_i$  and  $o_f = o_i$ . To have a transition we have to go from one state to a different state that is  $n_f \neq n_i$ . The transition is therefore

$$
D_{ox} = e \frac{2L^2}{\pi^2} \frac{n_i n_f [(-1)^{n_i - n_f} - 1]}{(n_i - n_f)^2 (n_i + n_f)^2} \frac{L}{2} \frac{L}{2} \quad \text{where } n_f \neq n_i.
$$

Now if  $n_i - n_f$  is an even number,  $[(-1)^{n_i - n_f} - 1]$  will vanish and  $D_{0x}$  will be zero. If  $n_i - n_f$ is an odd number,  $[(-1)^{n_i - n_f} - 1]$  will equal –2 and thus  $D_{0x}$  will be finite.

The *y*-component of the overlap integral will be

$$
D_{oy} = e^{\frac{L}{2}} \sin\left(\frac{n_i \pi x}{L}\right) \sin\left(\frac{n_j \pi x}{L}\right) dx \int_0^L y \sin\left(\frac{m_i \pi y}{L}\right) \sin\left(\frac{m_j \pi y}{L}\right) dy \int_0^L \sin\left(\frac{o_i \pi z}{L}\right) \sin\left(\frac{o_j \pi z}{L}\right) dz
$$

But this is identical to the expression for  $D_{0x}$  with *x* and *y* interchanged. Exactly the same occurs with *Doz*.

So it is clear that the only transitions that have a non-zero overlap, and hence emit a photon, are those where two quantum numbers remain unchanged and the third changes by an odd number.

The energy levels of the 3D symmetric potential well are given by  $E_{nmo} = E_1(n^2 + m^2 +$  $o<sup>2</sup>$ ), where  $E<sub>I</sub>$  is the ground state energy of the 1D infinite square well. If the transition is from *n* to  $n - p$ , the difference in energy is

$$
\Delta E = E_{nmo} - E_{(n-p)mo}
$$
  
=  $E_I(n^2 + m^2 + o^2) - E_I([n-p]^2 + m^2 + o^2)$   
=  $E_I(n^2 - [n-p]^2)$   
=  $p(2n-p) E_I$ 

2. For the hydrogen atom prove the selection rules ∆*m* = 0 or ±1 and ∆*l* = ±1. Remember, in spherical coordinates,  $\vec{r} = ir \sin \theta \sin \phi + jr \sin \theta \cos \phi + kr \cos \theta$ . Make use of symmetry where possible. The identities  $\int e^{-ib\phi} d\phi = \delta_{b0}$ 2 0  $\int_{0}^{2\pi} e^{-ib\phi} d\phi = \delta_{b0}$ ,  $\sin \phi = \frac{e^{i\phi} - e^{-i\phi}}{2i}$ 2  $\sin \phi = \frac{e^{i\phi} - e^{-i\phi}}{2i}$ φ  $=\frac{e^{i\phi}-e^{-i\phi}}{2i}$ , and  $\cos\phi=\frac{e^{i\phi}+2i}{2}$  $\cos \phi = \frac{e^{i\phi} + e^{-i\phi}}{2}$ φ  $=\frac{e^{i\phi}+e^{-i\phi}}{2}$ may be of use.

The wavefunctions for the hydrogen atom have the form  $\psi_{nlm} = AR_{nl}(r) f_{lm}(\theta) g_m(\phi)$ , where *A* is a normalization constant. The overlap integral indicates possible transitions

$$
\vec{D}_o = q \int \vec{r} \psi_i^* \psi_f dV.
$$

 Let the initial state be denoted by *nlm* and the final state by *qst*. Now note that the only imaginary term is  $g_m(\phi) = e^{im\phi}$  and its complex conjugate will be  $[g_m(\phi)]^* = e^{-im\phi} = g_{-m}(\phi)$ .

The overlap integral will be

$$
\vec{D}_0 = e \int_0^{\infty} \int_0^{\pi} \vec{r} R_{nl} f_{lm} g_{-m} R_{qs} f_{st} g_t r^2 \sin \theta dr d\theta d\phi
$$
\n
$$
= ie \int_0^{\infty} R_{nl} R_{qs} r^3 dr \int_0^{\pi} f_{lm} f_{st} \sin^2 \theta d\theta \int_0^{2\pi} g_{-m} g_t \sin \phi d\phi + i \int_0^{\infty} R_{nl} R_{qs} r^3 dr \int_0^{\pi} f_{lm} f_{st} \sin^2 \theta d\theta \int_0^{2\pi} g_{-m} g_t \cos \phi d\phi + i \int_0^{\infty} R_{nl} R_{qs} r^3 dr \int_0^{\pi} f_{lm} f_{st} \cos \theta \sin \theta d\theta \int_0^{2\pi} g_{-m} g_t d\phi
$$

Using the given identities, the  $\phi$  integrals become

$$
\int_{0}^{2\pi} g_{-m}g_{t} \sin \phi d\phi = \int_{0}^{2\pi} e^{-im\phi} e^{i\phi} \frac{e^{i\phi} - e^{-i\phi}}{2i} d\phi = \frac{-i}{2} \int_{0}^{2\pi} \left( e^{-i(m-t-1)} - e^{-i(m-t+1)} \right) d\phi = \frac{-i}{2} \left( \delta_{m,t+1} - \delta_{m,t-1} \right)
$$

and

$$
\int_{0}^{2\pi} g_{-m}g_{t} \cos \phi d\phi = \int_{0}^{2\pi} e^{-im\phi} e^{it\phi} \frac{e^{i\phi} + e^{-i\phi}}{2} d\phi = \frac{1}{2} \int_{0}^{2\pi} \left( e^{-i(m-t-1)} + e^{-i(m-t+1)} \right) d\phi = \frac{1}{2} \left( \delta_{m,t+1} + \delta_{m,t-1} \right)
$$

and

$$
\int_{0}^{2\pi} g_{-m} g_t d\phi = \int_{0}^{2\pi} e^{-im\phi} e^{it\phi} d\phi = \int_{0}^{2\pi} e^{-i(m-t)} d\phi = \delta_{m,t}.
$$

So we see that to get a non-zero contribution from the  $\phi$  integral we need  $t = m$  or  $t = m \pm 1$ 1, that is  $\Delta m = 0$  or  $\pm 1$ .

To determine the selection rule for *l*, we need to investigate the symmetry of *sin*(θ) and  $cos(\theta)$ , and  $f_{lm}(\theta)$  over the range  $0 \le \theta \le \pi$ .

First  $sin(\theta)$  is even and  $cos(\theta)$  is odd about the centre value  $\pi/2$ . Next consider the generating function for *flm*(θ),

$$
f_{lm}(\theta) = \frac{\sin^{|m|}(\theta)}{2^{\ell} \ell!} \left[ \frac{d}{d(\cos(\theta))} \right]^{\ell+|m|} (\cos^2(\theta)-1)^{\ell}.
$$

The  $sin^{10}(\theta)$  portion is even. The term  $(cos^2(\theta) - 1)^l$  only has even multiples of  $cos(\theta)$ , with leading term  $cos^{2l}(\theta)$ , so it is even. Applying the first derivative yields odd multiples of  $cos(\theta)$ , with leading term  $cos^{2l-1}(\theta)$ , so it is odd. The next derivative returns even multiples again, so overall it is even. The subsequent differentiations change from odd to even, and on and on. Therefore, if  $l + |m|$  is even then  $f_{lm}(\theta)$  is even, and if  $l + |m|$  is odd then  $f_{lm}(\theta)$  is odd. Also notice that the leading term after all the differentiations is  $cos^{2l - (l - 1)}$  $|m| \hat{\theta}| = \cos^{l-|m|}(\theta)$ . The term after that will be  $\cos^{l-|m|-2}(\theta)$  and so on.

Now consider the  $\theta$  integral of the  $D_{oz}$  term where we already know  $t = m$ ,

$$
\int\limits_{0}^{\pi} f_{lm} f_{sm} \cos \theta \sin \theta \, d\theta .
$$

The symmetry of this integral is determined by the leading term in the integrand,

$$
cos^{l-|m|}(\theta) cos^{s-|m|}(\theta) cos(\theta) sin^{2|m|+1}(\theta) = cos^{l+s+1-2|m|}(\theta) sin^{2|m|+1}(\theta).
$$

For this term to be even,  $l + s + 1 - 2|m|$  must be even. This demands that  $l + s$  be an odd number.

Now consider the  $\theta$  integral of the  $D_{ox}$  and  $D_{oy}$  terms where we already know  $t = m \pm 1$ ,

$$
\int\limits_{0}^{\pi} f_{l,m} f_{s,m\pm 1} \sin^2 \theta \, d\theta \, .
$$

The symmetry of this integral is determined by the leading term in the integrand,

$$
cos^{l-|m|}(\theta) cos^{s-|m\pm 1|}(\theta) sin^{2|m|+2}(\theta) = cos^{l+s-2|m|+1}(\theta) sin^{2|m|+2}(\theta).
$$

For this term to be even,  $l + s - 2|m| \pm 1$  must be even. This demands that  $l + s$  be an odd number.

So in each case we require  $\Delta l = \pm \text{odd}$ . We next need to now see why  $\Delta l = \pm 1$ .

To be continued!

3. Are  $L_x$ ,  $L_y$ ,  $L_z$  or  $L^2$  eigenfunctions of  $\psi(x)$  for the 3D potential well?

The wavefunction for the symmetric 3D potential is given by

$$
\psi(x, y, z) = \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi y}{L}\right) \sin\left(\frac{0\pi z}{L}\right).
$$

These operators are eigenfunctions of  $\psi(x, y, z)$  if, when they operate on  $\psi(x, y, z)$ , they yield a constant, i.e.  $O\psi = K\psi$ .

The first operator is  $L_x = y \left| \frac{n}{i} \frac{\partial}{\partial z} - z \right| = z \left| \frac{n}{i} \frac{\partial}{\partial y} \right|$  $\overline{\phantom{a}}$  $\lambda$  $\mathsf{I}$ l ſ ∂  $-\frac{1}{2}\left(\frac{\hbar}{2}\frac{\partial}{\partial z}\right)$  $\overline{1}$  $\left(\frac{\hbar}{\cdot}\frac{\partial}{\partial x}\right)^{x}$ l ſ  $= y \left( \frac{\hbar}{i} \frac{\partial}{\partial z} \right) - z \left( \frac{\hbar}{i} \frac{\partial}{\partial y} \right)$  $L_x = y \left( \frac{\hbar}{i} \frac{\partial}{\partial z} \right) - z \left( \frac{\hbar}{i} \frac{\partial}{\partial y} \right)$ . Operating on  $\psi(x, y, z)$  above yields

$$
L_x \psi = \sin\left(\frac{n\pi x}{L}\right) \left[\frac{\hbar}{i} \frac{\partial \pi y}{L} \sin\left(\frac{m\pi y}{L}\right) \cos\left(\frac{\partial \pi z}{L}\right) - \frac{\hbar}{i} \frac{m\pi z}{L} \cos\left(\frac{m\pi y}{L}\right) \sin\left(\frac{\partial \pi z}{L}\right)\right].
$$

This isn't an eigenfunction.

The second operator is  $L_v = z \frac{d^2 v}{dx^2} - x \frac{d^2 v}{dx^2}$  $\overline{\phantom{a}}$  $\left(\frac{\hbar}{\cdot}\frac{\partial}{\partial x}\right)^{n}$ l ſ ∂  $-\frac{x}{\sqrt{2}}\frac{\hbar}{\sqrt{2}}\frac{\partial}{\partial x}$  $\bigg)$  $\left(\frac{\hbar}{\cdot}\frac{\partial}{\partial x}\right)^{n}$ l ſ  $= z \left( \frac{\hbar}{i} \frac{\partial}{\partial x} \right) - x \left( \frac{\hbar}{i} \frac{\partial}{\partial z} \right)$  $L_y = z \left( \frac{\hbar}{i} \frac{\partial}{\partial x} \right) - x \left( \frac{\hbar}{i} \frac{\partial}{\partial z} \right)$ . Operating on  $\psi(x, y, z)$  above yields

$$
L_y \psi = \sin\left(\frac{m\pi y}{L}\right) \left[\frac{\hbar}{i} \frac{n\pi z}{L} \cos\left(\frac{n\pi x}{L}\right) \sin\left(\frac{\sigma \pi z}{L}\right) - \frac{\hbar}{i} \frac{\sigma \pi x}{L} \sin\left(\frac{n\pi x}{L}\right) \cos\left(\frac{\sigma \pi z}{L}\right)\right].
$$

This isn't an eigenfunction.

The third operator is  $L_z = x \left( \frac{h}{2} \right) - y \left( \frac{h}{2} \right)$  $\overline{1}$  $\left(\frac{\hbar}{\cdot}\frac{\partial}{\partial x}\right)^{x}$ l ſ  $-\frac{y}{i\frac{\hbar}{\partial x}}$  $\overline{\phantom{a}}$  $\overline{\phantom{a}}$  $\overline{\phantom{a}}$ l ſ  $L_z = x \left( \frac{\hbar}{i} \frac{\partial}{\partial y} \right) - y \left( \frac{\hbar}{i} \frac{\partial}{\partial x} \right)$ . Operating on  $\psi(x, y, z)$  above yields

$$
L_x \psi = \sin\left(\frac{0\pi z}{L}\right) \left[\frac{\hbar}{i} \frac{m\pi x}{L} \sin\left(\frac{n\pi x}{L}\right) \cos\left(\frac{m\pi y}{L}\right) - \frac{\hbar}{i} \frac{n\pi y}{L} \cos\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi y}{L}\right)\right].
$$

This isn't an eigenfunction.

Now  $L^2 = L_x^2 + L_y^2 + L_z^2$ , so we need to reapply the operators above. We find

$$
L_x(L_x \psi) = \hbar^2 \sin\left(\frac{n\pi x}{L}\right) \left\{ \left(\frac{o\pi y}{L}\right)^2 + \left(\frac{m\pi z}{L}\right)^2 \right\} \sin\left(\frac{m\pi y}{L}\right) \sin\left(\frac{o\pi z}{L}\right)
$$

$$
+ 2\frac{m\pi y}{L} \frac{o\pi z}{L} \cos\left(\frac{m\pi y}{L}\right) \cos\left(\frac{o\pi z}{L}\right)
$$

$$
+ \frac{m\pi y}{L} \cos\left(\frac{m\pi y}{L}\right) \sin\left(\frac{o\pi z}{L}\right) + \frac{o\pi z}{L} \sin\left(\frac{m\pi y}{L}\right) \cos\left(\frac{o\pi z}{L}\right)
$$

Similarly

$$
L_y(L_y \psi) = \hbar^2 \sin\left(\frac{m\pi y}{L}\right) \left\{ \left(\frac{0\pi x}{L}\right)^2 + \left(\frac{n\pi z}{L}\right)^2 \right\} \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{0\pi z}{L}\right)
$$

$$
+ 2\frac{n\pi x}{L} \frac{0\pi z}{L} \cos\left(\frac{m\pi y}{L}\right) \cos\left(\frac{0\pi z}{L}\right)
$$

$$
+ \frac{m\pi y}{L} \cos\left(\frac{m\pi y}{L}\right) \sin\left(\frac{0\pi z}{L}\right) + \frac{0\pi z}{L} \sin\left(\frac{m\pi y}{L}\right) \cos\left(\frac{0\pi z}{L}\right)
$$

and

$$
L_y(L_y \psi) = \hbar^2 \sin\left(\frac{\omega \pi z}{L}\right) \left\{ \left(\frac{m\pi x}{L}\right)^2 + \left(\frac{n\pi y}{L}\right)^2 \right\} \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi y}{L}\right)
$$

$$
+ 2\frac{n\pi x}{L} \frac{m\pi y}{L} \cos\left(\frac{n\pi x}{L}\right) \cos\left(\frac{m\pi y}{L}\right)
$$

$$
+ \frac{n\pi x}{L} \cos\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi y}{L}\right) + \frac{m\pi y}{L} \sin\left(\frac{n\pi x}{L}\right) \cos\left(\frac{m\pi y}{L}\right)
$$

Adding the above results, to get  $L^2 \psi = (L_x^2 + L_y^2 + L_z^2) \psi$  to does not reduce to a constant times  $\psi$  so this isn't an eigenfunction either.

4. Determine  $\langle r \rangle$ ,  $\langle r^2 \rangle$ , and  $\sigma_r$  for  $\psi_{100}$ . The integral identity  $\int x^n e^{-ax} dx = \frac{n!}{a^{n+1}}$ 0 ! + ∞  $\int x^n e^{-ax} dx = \frac{n}{a^n}$ *a*  $x^n e^{-ax} dx = \frac{n!}{n!}$  is helpful.

From the text,

$$
\Psi_{100}=R_{10}(r)Y_{00}(\theta,\phi)=\frac{2}{\sqrt{4\pi}}\sqrt{\left(\frac{Z}{a_0}\right)^3}e^{-Zr/a_0}.
$$

The expectation value of *r* is given by

$$
\langle r \rangle = \int r \psi^* \psi dV
$$
  
=  $\frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^3 \int_0^{\infty} r^3 e^{-2Zr/a_0} dr \int_0^{\pi} \sin^2 \theta d\theta \int_0^{2\pi} d\phi$   
=  $\frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^3 \int_0^{\infty} r^3 e^{-2Zr/a_0} dr \cdot 2 \cdot 2\pi$   
=  $4 \left( \frac{Z}{a_0} \right)^3 \frac{3!}{(2Z/a_0)^4}$   
=  $\frac{3}{2} \frac{a_0}{Z}$ 

Since  $Z = 1$  for hydrogen,  $\langle r \rangle = \frac{1}{2}a_0$ .

Similarly, the expectation value of  $r^2$  is given by

$$
\langle r^2 \rangle = \int r^2 \psi^* \psi dV
$$
  
=  $\frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^3 \int_0^{\infty} r^4 e^{-2Zr/a_0} dr \int_0^{\pi} \sin^2 \theta d\theta \int_0^{2\pi} d\phi$   
=  $\frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^3 \cdot \frac{4!}{(2Z/a_0)^5} \cdot 2 \cdot 2\pi$   
=  $3 \frac{a_0^2}{Z^2}$ 

Since  $Z = 1$  for hydrogen,  $\langle r^2 \rangle = 3a_0$ .

Thus 
$$
\sigma_r = \sqrt{r^2 - \overline{r}^2} = \sqrt{\frac{3}{4}} \frac{a_0}{Z}
$$
. Since  $Z = 1$  for hydrogen,  $\sigma_r = \sqrt{\frac{3}{4}} a_0$ .

5. Determine  $\langle p_r \rangle$ ,  $\langle p_r^2 \rangle$ , and  $\sigma_p$  for  $\psi_{100}$ . Note  $(p_r)$  $(p_r)_{op} = -i\hbar \frac{\partial}{\partial r}$  and  $(p_r^2)_{op} = -\hbar^2 \frac{1}{r^2} \frac{\partial}{\partial r} \left[ r^2 \frac{\partial}{\partial r} \right] = -\hbar^2 \left[ \frac{\partial}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r} \right]$  $\overline{\phantom{a}}$  $\overline{\phantom{a}}$  $\mathsf I$ l ſ ∂  $+\frac{2}{2}$  $\left(r^2 \frac{\partial}{\partial r}\right) = -\hbar^2 \left(\frac{\partial}{\partial r}\right)$ l ſ ∂ ∂  $=-\hbar^2 \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r}\right) = -\hbar^2 \left(\frac{\partial^2}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r}\right)$  $p_r^2\bigg|_{op} = -\hbar^2 \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r}\right) = -\hbar^2 \left(\frac{\partial^2}{\partial r^2} + \frac{2}{r}\right)$ 2  $\begin{pmatrix} 2 & \partial \end{pmatrix}$   $\begin{pmatrix} 2 & \lambda^2 \end{pmatrix}$  $\int_{r}^{2} \int_{op} = -\hbar^2 \frac{1}{r^2} \frac{\partial}{\partial r} \left[ r^2 \frac{\partial}{\partial r} \right] = -\hbar^2 \left[ \frac{\partial}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r} \right].$ 

From the previous question, 
$$
\Psi_{100} = \frac{2}{\sqrt{4\pi}} \sqrt{\left(\frac{Z}{a_0}\right)^3} e^{-Zr/a_0}
$$
. Thus  

$$
p_r \Psi_{100} = -i\hbar \frac{\partial}{\partial r} \Psi_{100} = i\hbar \frac{2}{\sqrt{4\pi}} \sqrt{\left(\frac{Z}{a_0}\right)^3} \frac{Z}{a_0} e^{-Zr/a_0}
$$

The expectation value of  $p_r$  is given by

$$
\langle p_r \rangle = \int \psi^* \left( -i\hbar \frac{\partial}{\partial r} \right) \psi dV
$$
  
=  $-i\hbar \frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^4 \int_0^{\infty} r^2 e^{-2Zr/a_0} dr \int_0^{\pi} \sin^2 \theta d\theta \int_0^{2\pi} d\phi$   
=  $-i\hbar \frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^4 \int_0^{\infty} r^2 e^{-2Zr/a_0} dr \cdot 2 \cdot 2\pi$   
=  $-i\hbar 4 \left( \frac{Z}{a_0} \right)^4 \frac{2!}{(2Z/a_0)^3}$   
=  $-i\hbar \frac{Z}{a_0}$ 

Since  $Z = 1$  for hydrogen,  $\langle p_r \rangle = -i\hbar/a_0$ .

Now 
$$
(p_r)^2 \psi_{100} = -\hbar^2 \left( \frac{\partial^2}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r} \right) \psi_{100} = -\hbar^2 \frac{2}{\sqrt{4\pi}} \left( \frac{Z}{a_0} \right)^{3/2} \left[ \left( \frac{Z}{a_0} \right)^2 - \frac{2}{r} \frac{Z}{a_0} \right] e^{-Zr/a_0}.
$$

The expectation value of  $p_r^2$  is given by

$$
\langle p_r^2 \rangle = \int \psi^* \hbar^2 \left( \frac{\partial^2}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r} \right) \psi dV
$$
  
=  $-\hbar^2 \frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^3 \int_0^{\infty} \left[ \left( \frac{Z}{a_0} \right)^2 - \frac{2}{r} \frac{Z}{a_0} \right] r^2 e^{-2Zr/a_0} dr \int_0^{\pi} \sin^2 \theta d\theta \int_0^{2\pi} d\phi$   
=  $-\hbar^2 \frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^3 \left[ \left( \frac{Z}{a_0} \right)^2 \frac{2!}{(2Z/a_0)^3} - 2 \frac{Z}{a_0} \frac{1!}{(2Z/a_0)^2} \right] \cdot 2 \cdot 2\pi$   
=  $\hbar^2 \left( \frac{Z}{a_0} \right)^2$ 

Thus 
$$
\sigma_p = \sqrt{p_r^2 - \overline{p}_r^2} = \hbar \sqrt{\left(\frac{Z}{a_0}\right)^2 - \left(\frac{Z}{a_0}\right)^2} = 0
$$
.

6. If electrons were spinning balls, they would have moment of inertia  $I = \frac{2}{5}MR^2$  and spin angular momentum  $S = I\omega$ . The maximum speed at which the outside edge of a real ball electron could be spinning is the speed of light. Since we know  $|\vec{S}| = \sqrt{s(s+1)}\hbar = \sqrt{\frac{3}{4}}\hbar$ , what value of R does this imply? Note that experiment has shown that electrons act like point particles down to distances below 10-15 m.

The speed of the outside edge is  $v = R\omega$ . Thus we have  $\frac{2}{5}MR^2 \frac{c}{R} = \sqrt{\frac{3}{4}}\hbar$  $\frac{2}{5}MR^2 \frac{c}{R} = \sqrt{\frac{3}{4}}\hbar$ . Solving for *R* 

we find  $R = \sqrt{\frac{3}{4} \frac{g}{g} \frac{m}{m^2}} = \sqrt{\frac{3}{4} \frac{g}{g} \frac{g}{g}} = \sqrt{3.5488888 \frac{g}{g}} = 8.36 \times 10^{-13}$  nm *eV eV nm Mc*  $R = \sqrt{\frac{3}{4} \frac{5}{2} \frac{hc}{Mc^2}} = \sqrt{\frac{3}{4} \frac{5}{2} \frac{197.33 \, eV \cdot nm}{0.5109989 \times 10^6 \, eV}} = 8.36 \times 10^{-13}$ 197.33 2 5 4 3 2 5 4 <sup>3</sup> <sup>−</sup> <sup>=</sup> <sup>×</sup>  $\times$  $=\sqrt{\frac{3}{4} \frac{5}{2} \frac{hc}{m}} = \sqrt{\frac{3}{4} \frac{5}{2} \frac{197.33 eV \cdot nm}{2 \cdot 51.00000} \cdot 10^{6} \cdot V}} = 8.36 \times 10^{-13} nm$ . Note that this

implies that the intrinsic quantum mechanical angular momentum called "*spin*" has no relation to the spin angular momentum of classical mechanics.

7. The deuteron is a hydrogen atom with an extra neutron in the core. The effects the reduced mass  $\mu$  in our equation for  $E_n$ . If  $m_p = 1836$   $m_e$  and  $m_n = 1839$   $m_e$ , find the wavelengths of light necessary to ionize a hydrogen atom and a deuteron. What is the difference in these two wavelengths?

 The equation for the energy levels of either version of hydrogen is given by 2  $\lambda^2$  $\frac{\mu}{2n^2} \left( \frac{n \Sigma c}{\hbar} \right)$  $\overline{1}$  $\overline{a}$  $\mathsf I$  $=\frac{\mu}{2n^2}\left(\frac{kZe}{\hbar}\right)$  $E_n = \frac{\mu}{2n^2} \left( \frac{kZe^2}{\hbar} \right)$  where  $Z = 1$ , *N e e m m m* + = 1  $\mu = \frac{m_e}{m}$ , and  $m_N$  is the mass of the nucleus. When

we deal just with hydrogen  $m_N = m_p$  and  $E_n^H = \frac{13.6}{\pi^2}$ *n*  $E_n^H = \frac{13.6 eV}{r^2}$ . For the deuteron,  $m_N = m_p + m_n$ ,

$$
E_n^D = \frac{\mu_D}{2n^2} \left(\frac{kZe^2}{\hbar}\right)^2 = \frac{\mu_D}{\mu_D} \frac{\mu}{2n^2} \left(\frac{kZe^2}{\hbar}\right)^2 = \frac{1 + \frac{m_e}{m_p + m_n}}{1 + \frac{m_e}{m_p}} \frac{13.6 \text{ eV}}{n^2}.
$$
 Using the given values

for the masses, we find  $E_n^D = 0.999728 \frac{13.6}{\pi^2}$ *n*  $E_n^D = 0.999728 \frac{13.6 eV}{n^2}$ . To ionize each atom we need  $E_1$ . The wavelength in each case is  $\lambda_H = \frac{hc}{\pi H} = \frac{1239.8 \text{ eV} \cdot nm}{24.62 \text{ eV}} = 91.162 \text{ eV}$ *E*  $h_H = \frac{hc}{E_1^H} = \frac{1239.8 \, eV \cdot nm}{13.6} = 91.162$ 1239.8 1  $\lambda_H = \frac{hc}{R_H} = \frac{1239.8 \, eV \cdot nm}{12.6 \, \epsilon} = 91.162 \, eV \, ,$ 

*eV eV eV nm E*  $\frac{hc}{E_1^D} = \frac{1239.8 \, eV \cdot nm}{(0.999728)(13.6 \, eV)} = 91.187$ 1239.8 1  $\lambda_D = \frac{hc}{\pi R} = \frac{1239.8 \text{ eV} \cdot nm}{(0.000720)(12.6 \text{ K})} = 91.187 \text{ eV}$ , and the difference is only 0.025 nm. 8. We can define the effective charge that an electron sees by the formula 2 2  $\omega_0$ *n E*  $E_n = Z_{\text{eff}}^2 \frac{L_0}{r^2}$  where  $E_0 = 13.6$  eV. The electron configuration of sodium is  $1s^2 2s^2 2p^6 3s$ . If the inner electrons fully shielded the valence electron from the core, what would *Z*eff be?

How much energy would be needed to remove the valence electron from sodium in this case? The actual ionization energy of sodium is 5.1 *eV*. What is the true *Z*eff?

From the electron configuration we see that the sodium valence electron has  $n = 3$ . If the valence electron was fully shielded then,  $E_3 = 1^2 (13.6 \text{ eV})/3^2 = 1.51 \text{ eV}$ . This is the amount of energy needed to remove the electron from the  $n = 3$  level. Knowing the correct value we see

$$
Z_{\text{eff}} = \sqrt{\frac{E_0 n^2}{E_n}} = \sqrt{\frac{5.1}{1.51}} = 1.84.
$$

9. Potassium has an electron configuration  $1s^2 2s^2 2p^6 3s^2 3p^6 4s$  and an ionization energy of 4.3 *eV*. What is the true *Z*eff? See previous question.

From the electron configuration we see that the potassium valence electron has  $n = 4$ . If Using the previous formula

$$
Z_{\text{eff}} = \sqrt{\frac{E_0 n^2}{E_n}} = \sqrt{\frac{(4.3)(16)}{13.6}} = 2.25.
$$

10. Write the electron configurations for the following, use only Fig. 7-19 of the text which shows the relative energies of the atomic shells and subshells. a) Rubidium,  $Z = 37$ . b) Iodine,  $Z = 53$ .

 We have to fill the orbitals shown in Fig. 7-19 starting at the lowest level and stopping when we run out of electrons. The order of filling and number of allowed electrons is 1*s*  $(2e)$ , 2s  $(2e)$ , 2p  $(6e)$ , 3s  $(2e)$ , 3p  $(6e)$ , 4s  $(2e)$ , 3d  $(10e)$ , 4p  $(6e)$ , 5s  $(2e)$ , 4d  $(10e)$ , 5p (6*e*), 6*s* (2*e*), 4*f* (*e*), 5*d* (14*e*), 6*p* (6*e*) …

Hence

- a) Rubidium  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$
- b) Iodine  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^5$