## 5 **Physics 2424**

1. The probability of spontaneous transitions of a charged particle from one initial energy level to another final level is governed by the overlap integral

$$\vec{D}_o = q \int \vec{r} \, \psi_i^* \psi_f dV \,,$$

where q is the charge of the particle. If  $\vec{D}_o = 0$  between two levels, the transition is forbidden. Searching for such forbidden transitions leads to *selection rules*. Consider the 3D infinite potential well.

- (a) Show that only allowed transitions from  $\psi_{n1,n2,n3}$  are to a state where two of the quantum numbers are unchanged and the other quantum number changes by an odd number like 1, 3, 5, etc.
- (b) Show that the emitted photons have energies  $E = p(2n p)E_1$  where the *n* is the original quantum level and n p is the final quantum level and  $E_1$  is the ground state energy of the one-dimensional potential well.

The following integral identities are useful:

$$\int_{0}^{L} \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi x}{L}\right) dx = \frac{L}{2} \delta_{nm} \text{ and } \int_{0}^{L} x \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi x}{L}\right) dx = \begin{cases} \frac{L^{2}}{4} & n = m \\ \frac{2L^{2}}{\pi^{2}} \frac{nm\left[(-1)^{n-m} - 1\right]}{(n-m)^{2}(n+m)^{2}} & n \neq m \end{cases}$$

The wavefunction for the symmetric 3D potential is given by

$$\psi(x, y, z) = \sin\left(\frac{n\pi x}{L}\right)\sin\left(\frac{m\pi y}{L}\right)\sin\left(\frac{o\pi z}{L}\right)$$

Also  $\vec{r} = ix + jy + kz$ . The overlap integral therefore is

$$\vec{D}_o = e \int_{0}^{L} \int_{0}^{L} \int_{0}^{L} (\mathbf{i}x + \mathbf{j}y + \mathbf{k}z) \sin\left(\frac{n_i \pi x}{L}\right) \sin\left(\frac{m_i \pi y}{L}\right) \sin\left(\frac{n_i \pi z}{L}\right) \sin\left(\frac{n_j \pi x}{L}\right) \sin\left(\frac{m_j \pi y}{L}\right) \sin\left(\frac{n_j \pi x}{L}\right) \sin\left($$

Considering the *x* component we find

$$D_{ox} = e \int_{0}^{L} x \sin\left(\frac{n_{i}\pi x}{L}\right) \sin\left(\frac{n_{f}\pi x}{L}\right) dx \int_{0}^{L} \sin\left(\frac{m_{i}\pi y}{L}\right) \sin\left(\frac{m_{f}\pi y}{L}\right) dy \int_{0}^{L} \sin\left(\frac{o_{i}\pi z}{L}\right) \sin\left(\frac{o_{f}\pi z}{L}\right) dz$$

The *y* and *z* integral will give zero unless  $m_f = m_i$  and  $o_f = o_i$ . To have a transition we have to go from one state to a different state that is  $n_f \neq n_i$ . The transition is therefore

$$D_{ox} = e \frac{2L^2}{\pi^2} \frac{n_i n_f \left[ (-1)^{n_i - n_f} - 1 \right]}{(n_i - n_f)^2 (n_i + n_f)^2} \frac{L}{2} \frac{L}{2} \quad where \ n_f \neq n_i.$$

Now if  $n_i - n_f$  is an even number,  $[(-1)^{n_i - n_f} - 1]$  will vanish and  $D_{0x}$  will be zero. If  $n_i - n_f$  is an odd number,  $[(-1)^{n_i - n_f} - 1]$  will equal -2 and thus  $D_{0x}$  will be finite.

The y-component of the overlap integral will be

$$D_{oy} = e_0^L \sin\left(\frac{n_i \pi x}{L}\right) \sin\left(\frac{n_f \pi x}{L}\right) dx_0^L y \sin\left(\frac{m_i \pi y}{L}\right) \sin\left(\frac{m_f \pi y}{L}\right) dy_0^L \sin\left(\frac{o_i \pi z}{L}\right) \sin\left(\frac{o_f \pi z}{L}\right) dz$$

But this is identical to the expression for  $D_{0x}$  with x and y interchanged. Exactly the same occurs with  $D_{oz}$ .

So it is clear that the only transitions that have a non-zero overlap, and hence emit a photon, are those where two quantum numbers remain unchanged and the third changes by an odd number.

The energy levels of the 3D symmetric potential well are given by  $E_{nmo} = E_1(n^2 + m^2 + o^2)$ , where  $E_1$  is the ground state energy of the 1D infinite square well. If the transition is from *n* to n - p, the difference in energy is

$$\Delta E = E_{nmo} - E_{(n-p)mo}$$
  
=  $E_I(n^2 + m^2 + o^2) - E_I([n-p]^2 + m^2 + o^2)$   
=  $E_I(n^2 - [n-p]^2)$   
=  $p(2n-p) E_I$ 

2. For the hydrogen atom prove the selection rules  $\Delta m = 0$  or  $\pm 1$  and  $\Delta l = \pm 1$ . Remember, in spherical coordinates,  $\vec{r} = ir \sin \theta \sin \phi + jr \sin \theta \cos \phi + kr \cos \theta$ . Make use of symmetry where possible. The identities  $\int_{0}^{2\pi} e^{-ib\phi} d\phi = \delta_{b0}$ ,  $\sin \phi = \frac{e^{i\phi} - e^{-i\phi}}{2i}$ , and  $\cos \phi = \frac{e^{i\phi} + e^{-i\phi}}{2}$  may be of use.

The wavefunctions for the hydrogen atom have the form  $\psi_{nlm} = AR_{nl}(r)f_{lm}(\theta)g_m(\phi)$ , where *A* is a normalization constant. The overlap integral indicates possible transitions

$$\vec{D}_o = q \int \vec{r} \psi_i^* \psi_f dV \,.$$

Let the initial state be denoted by *nlm* and the final state by *qst*. Now note that the only imaginary term is  $g_m(\phi) = e^{im\phi}$  and its complex conjugate will be  $[g_m(\phi)]^* = e^{-im\phi} = g_{-m}(\phi)$ .

The overlap integral will be

$$\vec{D}_{0} = e \int_{0}^{\infty} \int_{0}^{\pi} \int_{0}^{2\pi} \vec{r} R_{nl} f_{lm} g_{-m} R_{qs} f_{st} g_{t} r^{2} \sin\theta dr d\vartheta d\phi$$

$$= i e \int_{0}^{\infty} R_{nl} R_{qs} r^{3} dr \int_{0}^{\pi} f_{lm} f_{st} \sin^{2}\theta d\theta \int_{0}^{2\pi} g_{-m} g_{t} \sin\phi d\phi +$$

$$j e \int_{0}^{\infty} R_{nl} R_{qs} r^{3} dr \int_{0}^{\pi} f_{lm} f_{st} \sin^{2}\theta d\theta \int_{0}^{2\pi} g_{-m} g_{t} \cos\phi d\phi +$$

$$k e \int_{0}^{\infty} R_{nl} R_{qs} r^{3} dr \int_{0}^{\pi} f_{lm} f_{st} \cos\theta \sin\theta d\theta \int_{0}^{2\pi} g_{-m} g_{t} d\phi \quad .$$

Using the given identities, the  $\phi$  integrals become

$$\int_{0}^{2\pi} g_{-m}g_{t}\sin\phi d\phi = \int_{0}^{2\pi} e^{-im\phi}e^{it\phi} \frac{e^{i\phi} - e^{-i\phi}}{2i}d\phi = \frac{-i}{2}\int_{0}^{2\pi} \left(e^{-i(m-t-1)} - e^{-i(m-t+1)}\right)d\phi = \frac{-i}{2}\left(\delta_{m,t+1} - \delta_{m,t-1}\right)$$

and

$$\int_{0}^{2\pi} g_{-m}g_{t}\cos\phi d\phi = \int_{0}^{2\pi} e^{-im\phi}e^{it\phi} \frac{e^{i\phi} + e^{-i\phi}}{2}d\phi = \frac{1}{2}\int_{0}^{2\pi} \left(e^{-i(m-t-1)} + e^{-i(m-t+1)}\right)d\phi = \frac{1}{2}\left(\delta_{m,t+1} + \delta_{m,t-1}\right)$$

and

$$\int_{0}^{2\pi} g_{-m} g_{t} d\phi = \int_{0}^{2\pi} e^{-im\phi} e^{it\phi} d\phi = \int_{0}^{2\pi} e^{-i(m-t)} d\phi = \delta_{m,t}.$$

So we see that to get a non-zero contribution from the  $\phi$  integral we need t = m or  $t = m \pm 1$ , that is  $\Delta m = 0$  or  $\pm 1$ .

To determine the selection rule for *l*, we need to investigate the symmetry of  $sin(\theta)$  and  $cos(\theta)$ , and  $f_{lm}(\theta)$  over the range  $0 \le \theta \le \pi$ .

First  $sin(\theta)$  is even and  $cos(\theta)$  is odd about the centre value  $\pi/2$ . Next consider the generating function for  $f_{lm}(\theta)$ ,

$$f_{lm}(\theta) = \frac{\sin^{|m|}(\theta)}{2^{\ell} \ell!} \left[ \frac{d}{d(\cos(\theta))} \right]^{\ell+|m|} \left( \cos^2(\theta) - 1 \right)^{\ell}.$$

The  $sin^{|m|}(\theta)$  portion is even. The term  $(cos^2(\theta) - 1)^l$  only has even multiples of  $cos(\theta)$ , with leading term  $cos^{2l}(\theta)$ , so it is even. Applying the first derivative yields odd multiples of  $cos(\theta)$ , with leading term  $cos^{2l-1}(\theta)$ , so it is odd. The next derivative returns even multiples again, so overall it is even. The subsequent differentiations change from odd to even, and on and on. Therefore, if l + |m| is even then  $f_{lm}(\theta)$  is even, and if l + |m| is odd then  $f_{lm}(\theta)$  is odd. Also notice that the leading term after all the differentiations is  $cos^{2l-(l-1)}(\theta) = cos^{l-|m|}(\theta)$ . The term after that will be  $cos^{l-|m|-2}(\theta)$  and so on.

Now consider the  $\theta$  integral of the  $D_{oz}$  term where we already know t = m,

$$\int_{0}^{\pi} f_{lm} f_{sm} \cos \theta \sin \theta \, d\theta \, .$$

The symmetry of this integral is determined by the leading term in the integrand,

$$\cos^{l-|m|}(\theta)\cos^{s-|m|}(\theta)\cos(\theta)\sin^{2|m|+1}(\theta) = \cos^{l+s+1-2|m|}(\theta)\sin^{2|m|+1}(\theta)$$

For this term to be even, l + s + 1 - 2|m| must be even. This demands that l + s be an odd number.

Now consider the  $\theta$  integral of the  $D_{ox}$  and  $D_{oy}$  terms where we already know  $t = m \pm 1$ ,

$$\int_{0}^{\pi} f_{l,m} f_{s,m\pm 1} \sin^2 \theta \, d\theta \, .$$

The symmetry of this integral is determined by the leading term in the integrand,

$$\cos^{l-|m|}(\theta)\cos^{s-|m\pm 1|}(\theta)\sin^{2|m|+2}(\theta) = \cos^{l+s-2|m|\pm 1}(\theta)\sin^{2|m|+2}(\theta)$$

For this term to be even,  $l + s - 2|m| \pm 1$  must be even. This demands that l + s be an odd number.

So in each case we require  $\Delta l = \pm \text{odd}$ . We next need to now see why  $\Delta l = \pm 1$ .

To be continued!

3. Are  $L_x$ ,  $L_y$ ,  $L_z$  or  $L^2$  eigenfunctions of  $\psi(x)$  for the 3D potential well?

The wavefunction for the symmetric 3D potential is given by

$$\Psi(x, y, z) = \sin\left(\frac{n\pi x}{L}\right)\sin\left(\frac{m\pi y}{L}\right)\sin\left(\frac{o\pi z}{L}\right).$$

These operators are eigenfunctions of  $\psi(x, y, z)$  if, when they operate on  $\psi(x, y, z)$ , they yield a constant, i.e.  $O\psi = K\psi$ .

The first operator is  $L_x = y \left(\frac{\hbar}{i} \frac{\partial}{\partial z}\right) - z \left(\frac{\hbar}{i} \frac{\partial}{\partial y}\right)$ . Operating on  $\psi(x, y, z)$  above yields

$$L_x \psi = \sin\left(\frac{n\pi x}{L}\right) \left[\frac{\hbar}{i} \frac{o\pi y}{L} \sin\left(\frac{m\pi y}{L}\right) \cos\left(\frac{o\pi z}{L}\right) - \frac{\hbar}{i} \frac{m\pi z}{L} \cos\left(\frac{m\pi y}{L}\right) \sin\left(\frac{o\pi z}{L}\right)\right].$$

This isn't an eigenfunction.

The second operator is  $L_y = z \left(\frac{\hbar}{i} \frac{\partial}{\partial x}\right) - x \left(\frac{\hbar}{i} \frac{\partial}{\partial z}\right)$ . Operating on  $\psi(x, y, z)$  above yields

$$L_{y}\psi = \sin\left(\frac{m\pi y}{L}\right)\left[\frac{\hbar}{i}\frac{n\pi z}{L}\cos\left(\frac{n\pi x}{L}\right)\sin\left(\frac{o\pi z}{L}\right) - \frac{\hbar}{i}\frac{o\pi x}{L}\sin\left(\frac{n\pi x}{L}\right)\cos\left(\frac{o\pi z}{L}\right)\right].$$

This isn't an eigenfunction.

The third operator is  $L_z = x \left( \frac{\hbar}{i} \frac{\partial}{\partial y} \right) - y \left( \frac{\hbar}{i} \frac{\partial}{\partial x} \right)$ . Operating on  $\psi(x, y, z)$  above yields

$$L_x \psi = \sin\left(\frac{\sigma\pi z}{L}\right) \left[\frac{\hbar}{i} \frac{m\pi x}{L} \sin\left(\frac{n\pi x}{L}\right) \cos\left(\frac{m\pi y}{L}\right) - \frac{\hbar}{i} \frac{n\pi y}{L} \cos\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi y}{L}\right) \right].$$

This isn't an eigenfunction.

Now  $L^2 = L_x^2 + L_y^2 + L_z^2$ , so we need to reapply the operators above. We find

$$L_{x}(L_{x}\psi) = \hbar^{2} \sin\left(\frac{n\pi x}{L}\right) \left\{ \left\{ \left(\frac{o\pi y}{L}\right)^{2} + \left(\frac{m\pi z}{L}\right)^{2} \right\} \sin\left(\frac{m\pi y}{L}\right) \sin\left(\frac{o\pi z}{L}\right) + 2\frac{m\pi y}{L}\frac{o\pi z}{L}\cos\left(\frac{m\pi y}{L}\right) \cos\left(\frac{o\pi z}{L}\right) + \frac{m\pi y}{L}\cos\left(\frac{m\pi y}{L}\right) \sin\left(\frac{o\pi z}{L}\right) + \frac{o\pi z}{L}\sin\left(\frac{m\pi y}{L}\right) \cos\left(\frac{o\pi z}{L}\right) \right\}$$

Similarly

$$L_{y}(L_{y}\psi) = \hbar^{2} \sin\left(\frac{m\pi y}{L}\right) \left[ \left\{ \left(\frac{o\pi x}{L}\right)^{2} + \left(\frac{n\pi z}{L}\right)^{2} \right\} \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{o\pi z}{L}\right) \right. \\ \left. + 2\frac{n\pi x}{L}\frac{o\pi z}{L}\cos\left(\frac{m\pi y}{L}\right) \cos\left(\frac{o\pi z}{L}\right) \right] \\ \left. + \frac{m\pi y}{L}\cos\left(\frac{m\pi y}{L}\right) \sin\left(\frac{o\pi z}{L}\right) + \frac{o\pi z}{L}\sin\left(\frac{m\pi y}{L}\right) \cos\left(\frac{o\pi z}{L}\right) \right]$$

and

$$L_{y}(L_{y}\psi) = \hbar^{2} \sin\left(\frac{n\pi z}{L}\right) \left[ \left\{ \left(\frac{m\pi x}{L}\right)^{2} + \left(\frac{n\pi y}{L}\right)^{2} \right\} \sin\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi y}{L}\right) + 2\frac{n\pi x}{L}\frac{m\pi y}{L}\cos\left(\frac{n\pi x}{L}\right) \cos\left(\frac{m\pi y}{L}\right) + \frac{n\pi x}{L}\cos\left(\frac{n\pi x}{L}\right) \sin\left(\frac{m\pi y}{L}\right) + \frac{m\pi y}{L}\sin\left(\frac{n\pi x}{L}\right) \cos\left(\frac{m\pi y}{L}\right) \right]$$

Adding the above results, to get  $L^2 \psi = (L_x^2 + L_y^2 + L_z^2) \psi$  to does not reduce to a constant times  $\psi$  so this isn't an eigenfunction either.

4. Determine  $\langle r \rangle$ ,  $\langle r^2 \rangle$ , and  $\sigma_r$  for  $\psi_{100}$ . The integral identity  $\int_{0}^{\infty} x^n e^{-ax} dx = \frac{n!}{a^{n+1}}$  is helpful.

From the text,

$$\psi_{100} = R_{10}(r)Y_{00}(\theta,\phi) = \frac{2}{\sqrt{4\pi}}\sqrt{\left(\frac{Z}{a_0}\right)^3}e^{-Zr/a_0}.$$

The expectation value of r is given by

$$\langle r \rangle = \int r \psi^* \psi dV$$
  
=  $\frac{4}{4\pi} \left(\frac{Z}{a_0}\right)^3 \int_0^\infty r^3 e^{-2Zr/a_0} dr \int_0^\pi \sin^2 \theta \, d\theta \int_0^{2\pi} d\phi$   
=  $\frac{4}{4\pi} \left(\frac{Z}{a_0}\right)^3 \int_0^\infty r^3 e^{-2Zr/a_0} dr \cdot 2 \cdot 2\pi$   
=  $4 \left(\frac{Z}{a_0}\right)^3 \frac{3!}{(2Z/a_0)^4}$   
=  $\frac{3}{2} \frac{a_0}{Z}$ 

Since Z = 1 for hydrogen,  $\langle r \rangle = \frac{1}{2}a_0$ .

Similarly, the expectation value of  $r^2$  is given by

$$\left\langle r^{2} \right\rangle = \int r^{2} \psi^{*} \psi dV$$
$$= \frac{4}{4\pi} \left( \frac{Z}{a_{0}} \right)^{3} \int_{0}^{\infty} r^{4} e^{-2Zr/a_{0}} dr \int_{0}^{\pi} \sin^{2} \theta \, d\theta \int_{0}^{2\pi} d\phi$$
$$= \frac{4}{4\pi} \left( \frac{Z}{a_{0}} \right)^{3} \cdot \frac{4!}{\left( \frac{2Z}{a_{0}} \right)^{5}} \cdot 2 \cdot 2\pi$$
$$= 3 \frac{a_{0}^{2}}{Z^{2}}$$

Since Z = 1 for hydrogen,  $\langle r^2 \rangle = 3a_0$ .

Thus 
$$\sigma_r = \sqrt{r^2 - \overline{r}^2} = \sqrt{\frac{3}{4}} \frac{a_0}{Z}$$
. Since  $Z = 1$  for hydrogen,  $\sigma_r = \sqrt{\frac{3}{4}} a_0$ .

5. Determine  $\langle p_r \rangle$ ,  $\langle p_r^2 \rangle$ , and  $\sigma_p$  for  $\psi_{100}$ . Note  $(p_r)_{op} = -i\hbar \frac{\partial}{\partial r}$  and  $(p_r^2)_{op} = -\hbar^2 \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r}\right) = -\hbar^2 \left(\frac{\partial^2}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r}\right).$ 

From the previous question, 
$$\psi_{100} = \frac{2}{\sqrt{4\pi}} \sqrt{\left(\frac{Z}{a_0}\right)^3} e^{-Zr/a_0}$$
. Thus  
 $p_r \psi_{100} = -i\hbar \frac{\partial}{\partial r} \psi_{100} = i\hbar \frac{2}{\sqrt{4\pi}} \sqrt{\left(\frac{Z}{a_0}\right)^3} \frac{Z}{a_0} e^{-Zr/a_0}$ 

The expectation value of  $p_r$  is given by

$$\langle p_r \rangle = \int \psi^* \left( -i\hbar \frac{\partial}{\partial r} \right) \psi dV$$

$$= -i\hbar \frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^4 \int_0^\infty r^2 e^{-2Zr/a_0} dr \int_0^\pi \sin^2 \theta \, d\theta \int_0^{2\pi} d\phi$$

$$= -i\hbar \frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^4 \int_0^\infty r^2 e^{-2Zr/a_0} dr \cdot 2 \cdot 2\pi$$

$$= -i\hbar 4 \left( \frac{Z}{a_0} \right)^4 \frac{2!}{(2Z/a_0)^3}$$

$$= -i\hbar \frac{Z}{a_0}$$

Since Z = 1 for hydrogen,  $\langle p_r \rangle = -i\hbar/a_0$ .

$$\operatorname{Now}(p_r)^2 \psi_{100} = -\hbar^2 \left( \frac{\partial^2}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r} \right) \psi_{100} = -\hbar^2 \frac{2}{\sqrt{4\pi}} \left( \frac{Z}{a_0} \right)^{3/2} \left[ \left( \frac{Z}{a_0} \right)^2 - \frac{2}{r} \frac{Z}{a_0} \right] e^{-Zr/a_0}.$$

The expectation value of  $p_r^2$  is given by

$$\left\langle p_r^2 \right\rangle = \int \psi^* \hbar^2 \left( \frac{\partial^2}{\partial r^2} + \frac{2}{r} \frac{\partial}{\partial r} \right) \psi dV$$

$$= -\hbar^2 \frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^3 \int_0^\infty \left[ \left( \frac{Z}{a_0} \right)^2 - \frac{2}{r} \frac{Z}{a_0} \right] r^2 e^{-2Zr/a_0} dr \int_0^\pi \sin^2 \theta \, d\theta \int_0^{2\pi} d\phi$$

$$= -\hbar^2 \frac{4}{4\pi} \left( \frac{Z}{a_0} \right)^3 \left[ \left( \frac{Z}{a_0} \right)^2 \frac{2!}{(2Z/a_0)^3} - 2\frac{Z}{a_0} \frac{1!}{(2Z/a_0)^2} \right] \cdot 2 \cdot 2\pi$$

$$= \hbar^2 \left( \frac{Z}{a_0} \right)^2$$

Thus 
$$\sigma_p = \sqrt{p_r^2 - \overline{p}_r^2} = \hbar \sqrt{\left(\frac{Z}{a_0}\right)^2 - \left(\frac{Z}{a_0}\right)^2} = 0$$
.

6. If electrons were spinning balls, they would have moment of inertia  $I = \frac{2}{5}MR^2$  and spin angular momentum  $S = I\omega$ . The maximum speed at which the outside edge of a real ball electron could be spinning is the speed of light. Since we know  $\left|\vec{S}\right| = \sqrt{s(s+1)}\hbar = \sqrt{\frac{3}{4}}\hbar$ , what value of R does this imply? Note that experiment has shown that electrons act like point particles down to distances below 10<sup>-15</sup> m.

The speed of the outside edge is  $v = R\omega$ . Thus we have  $\frac{2}{5}MR^2\frac{c}{R} = \sqrt{\frac{3}{4}}\hbar$ . Solving for R

we find  $R = \sqrt{\frac{3}{4}} \frac{5}{2} \frac{\hbar c}{Mc^2} = \sqrt{\frac{3}{4}} \frac{5}{2} \frac{197.33 \, eV \cdot nm}{0.5109989 \times 10^6 \, eV} = 8.36 \times 10^{-13} \, nm$ . Note that this

implies that the intrinsic quantum mechanical angular momentum called "*spin*" has no relation to the spin angular momentum of classical mechanics.

7. The deuteron is a hydrogen atom with an extra neutron in the core. The effects the reduced mass  $\mu$  in our equation for  $E_n$ . If  $m_p = 1836 m_e$  and  $m_n = 1839 m_e$ , find the wavelengths of light necessary to ionize a hydrogen atom and a deuteron. What is the difference in these two wavelengths?

The equation for the energy levels of either version of hydrogen is given by  $E_n = \frac{\mu}{2n^2} \left(\frac{kZe^2}{\hbar}\right)^2 \text{ where } Z = 1, \ \mu = \frac{m_e}{1 + \frac{m_e}{m_N}}, \text{ and } m_N \text{ is the mass of the nucleus. When}$ 

we deal just with hydrogen  $m_N = m_p$  and  $E_n^H = \frac{13.6 eV}{n^2}$ . For the deuteron,  $m_N = m_p + m_n$ ,

$$E_n^D = \frac{\mu_D}{2n^2} \left(\frac{kZe^2}{\hbar}\right)^2 = \frac{\mu_D}{\mu_D} \frac{\mu}{2n^2} \left(\frac{kZe^2}{\hbar}\right)^2 = \frac{1 + \frac{m_e}{m_p + m_n}}{1 + \frac{m_e}{m_p}} \frac{13.6 \, eV}{n^2}.$$
 Using the given values

for the masses, we find  $E_n^D = 0.999728 \frac{13.6 \, eV}{n^2}$ . To ionize each atom we need  $E_1$ . The wavelength in each case is  $\lambda_H = \frac{hc}{E_1^H} = \frac{1239.8 \, eV \cdot nm}{13.6} = 91.162 \, eV$ ,

 $\lambda_D = \frac{hc}{E_1^D} = \frac{1239.8 \, eV \cdot nm}{(0.999728)(13.6 \, eV)} = 91.187 \, eV$ , and the difference is only 0.025 nm.

8. We can define the effective charge that an electron sees by the formula  $E_n = Z_{eff}^2 \frac{E_0}{n^2}$  where  $E_0 = 13.6$  eV. The electron configuration of sodium is  $1s^2 2s^2 2p^6 3s$ . If
the inner electrons fully shielded the valence electron from the core, what would  $Z_{eff}$  be?

the inner electrons fully shielded the valence electron from the core, what would  $Z_{eff}$  be? How much energy would be needed to remove the valence electron from sodium in this case? The actual ionization energy of sodium is 5.1 *eV*. What is the true  $Z_{eff}$ ?

From the electron configuration we see that the sodium valence electron has n = 3. If the valence electron was fully shielded then,  $E_3 = 1^2 (13.6 \text{ eV})/3^2 = 1.51 \text{ eV}$ . This is the amount of energy needed to remove the electron from the n = 3 level. Knowing the correct value we see

$$Z_{eff} = \sqrt{\frac{E_0 n^2}{E_n}} = \sqrt{\frac{5.1}{1.51}} = 1.84$$

9. Potassium has an electron configuration  $1s^22s^22p^63s^23p^64s$  and an ionization energy of 4.3 *eV*. What is the true  $Z_{\text{eff}}$ ? See previous question.

From the electron configuration we see that the potassium valence electron has n = 4. If Using the previous formula

$$Z_{eff} = \sqrt{\frac{E_0 n^2}{E_n}} = \sqrt{\frac{(4.3)(16)}{13.6}} = 2.25$$
.

10. Write the electron configurations for the following, use only Fig. 7-19 of the text which shows the relative energies of the atomic shells and subshells.
a) Rubidium, Z = 37.
b) Iodine, Z = 53.

We have to fill the orbitals shown in Fig. 7-19 starting at the lowest level and stopping when we run out of electrons. The order of filling and number of allowed electrons is 1s (2e), 2s (2e), 2p (6e), 3s (2e), 3p (6e), 4s (2e), 3d (10e), 4p (6e), 5s (2e), 4d (10e), 5p (6e), 6s (2e), 4f(e), 5d(14e), 6p(6e) ...

Hence

- a) Rubidium  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$
- b) Iodine  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^5$